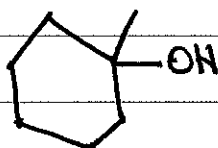


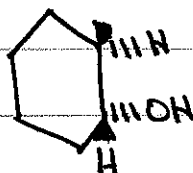
* Rxns of Alcohols Application Answers

①

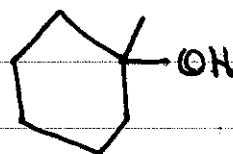
a.)



Hg

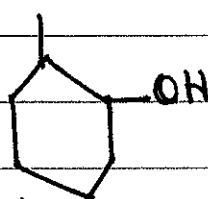


BH₃

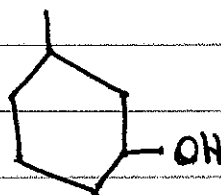


H₂O

b.)



+

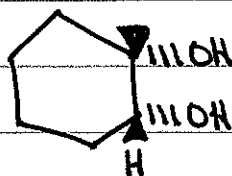


Hg + BH₃

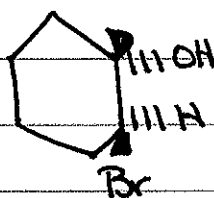


H₂O

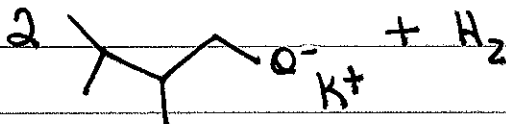
c.)



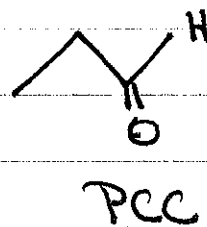
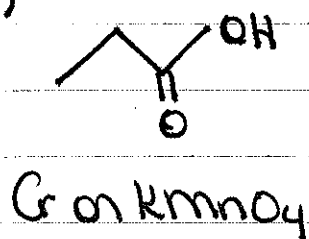
d.)



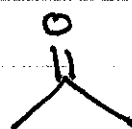
e.)



f.)



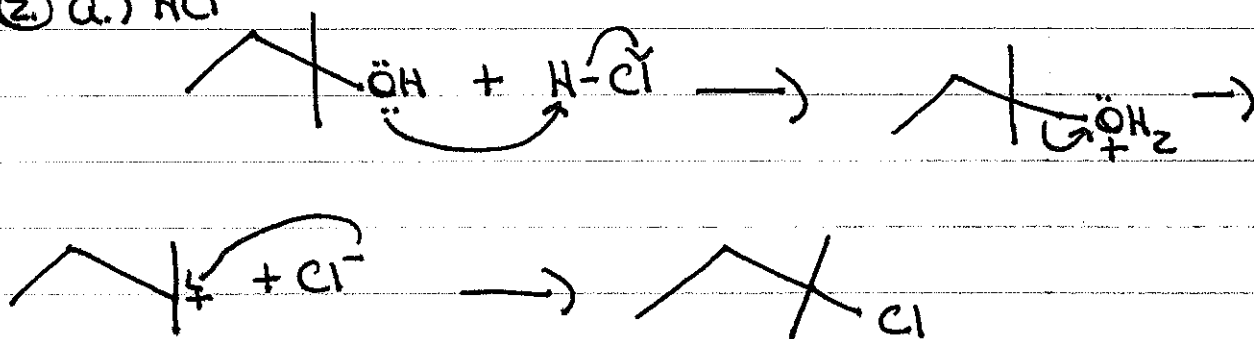
g.



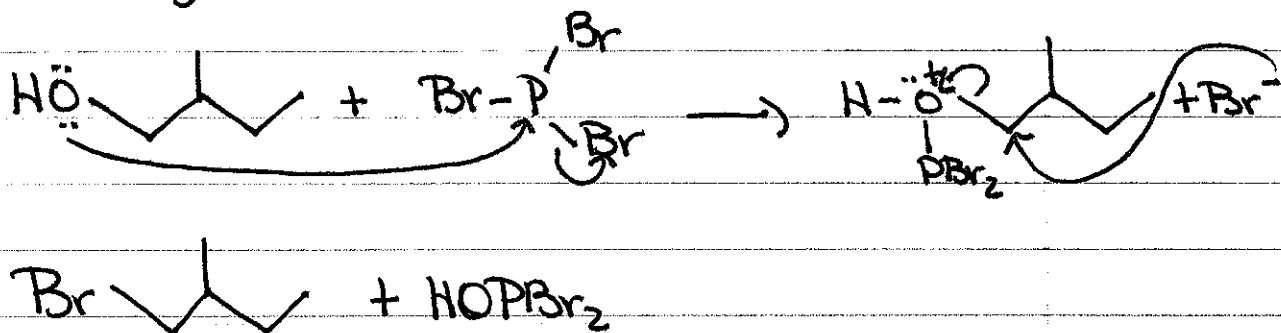
Cr, KMnO_4 , PCC

h.) No Rxn Never oxidize 3° ROH!

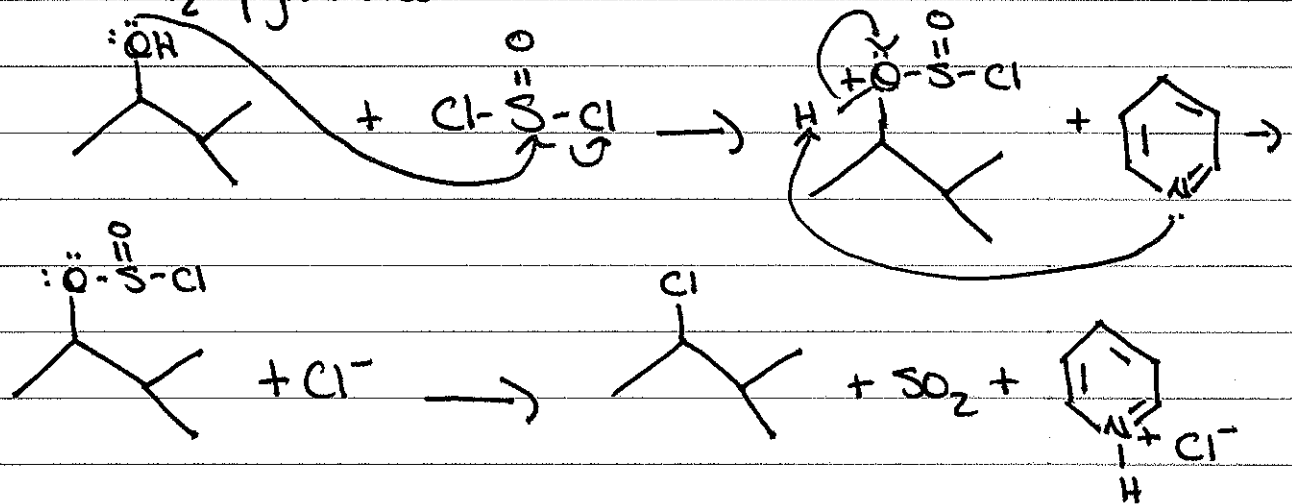
② a.) HCl



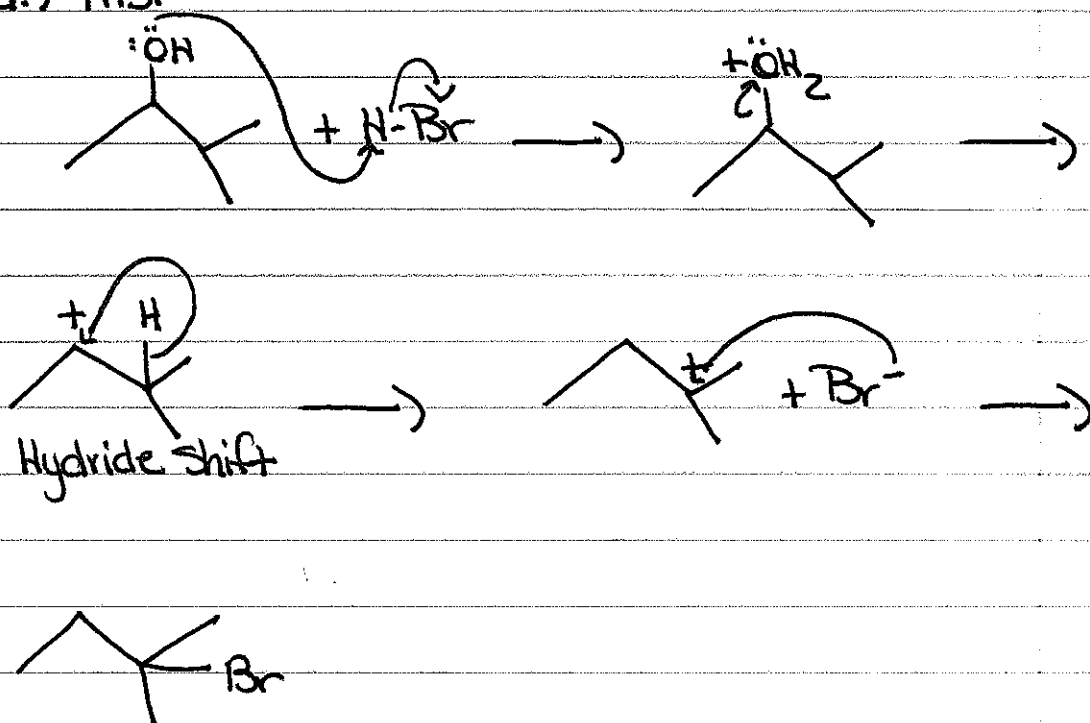
b.) PBr_3 + Ether



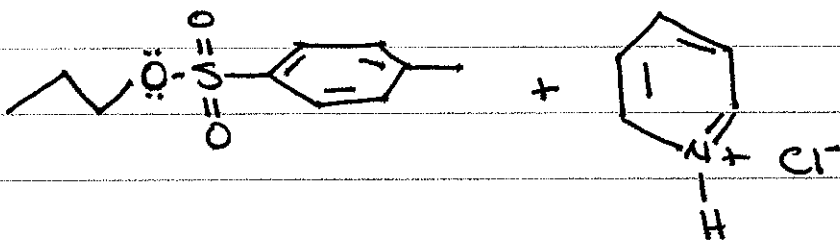
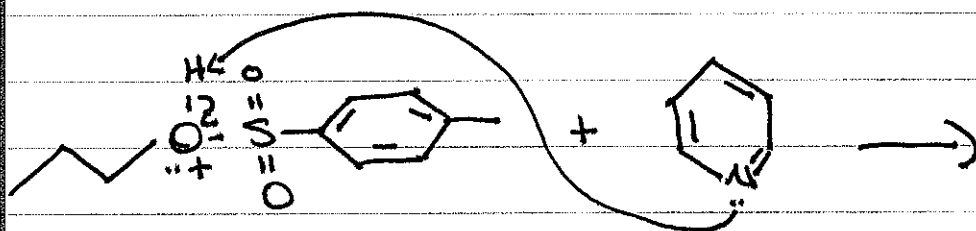
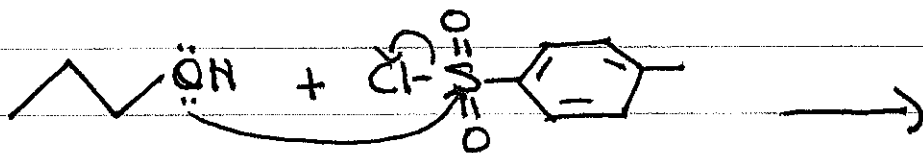
c.) $\text{SOCl}_2 + \text{pyridine}$



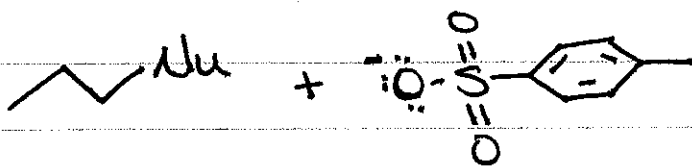
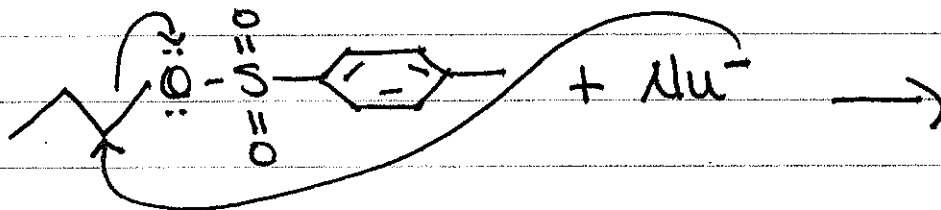
d.) HBr

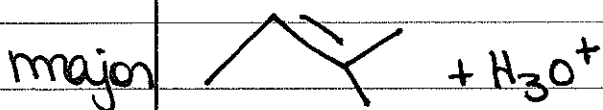
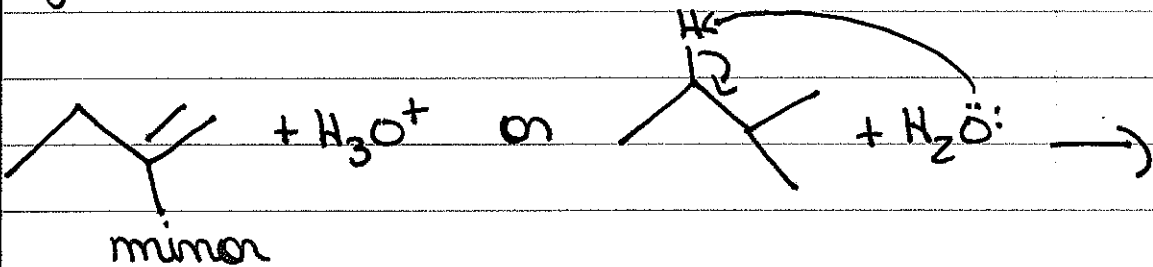
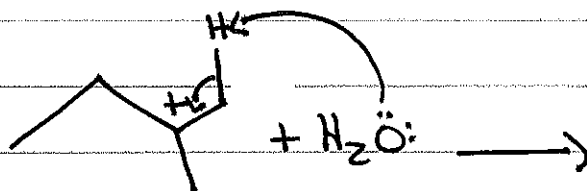
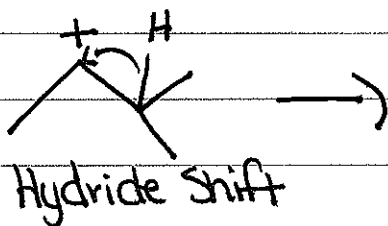
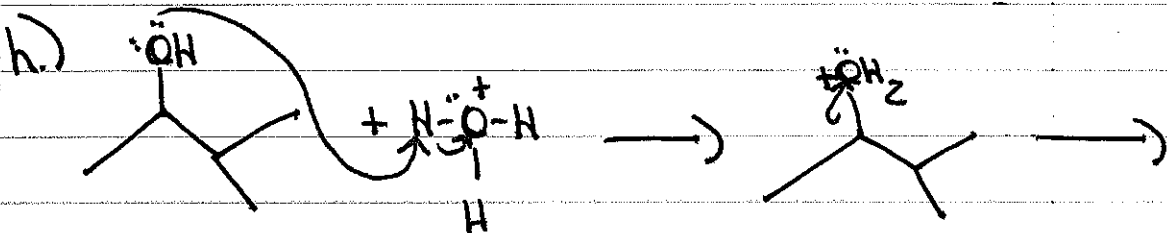
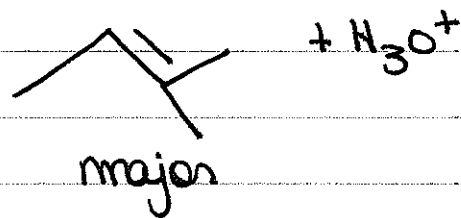
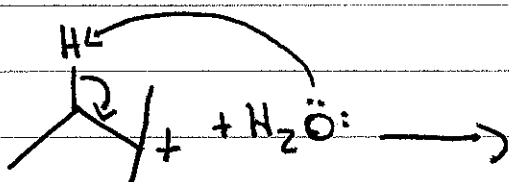
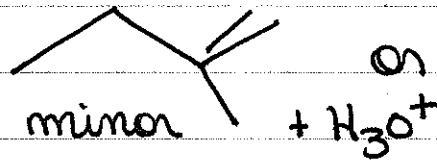
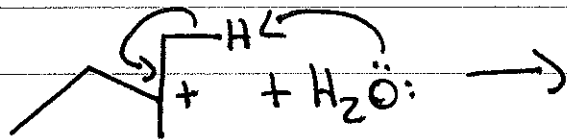
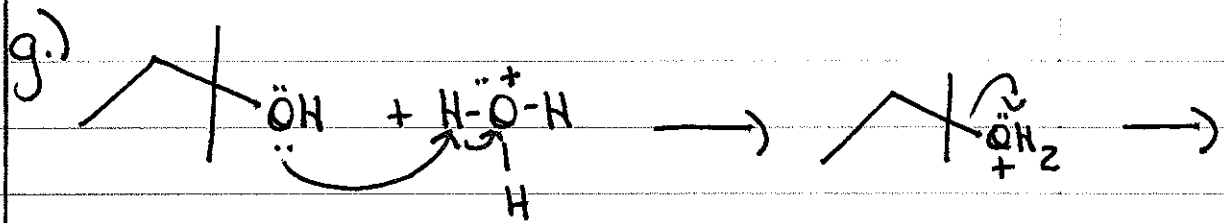


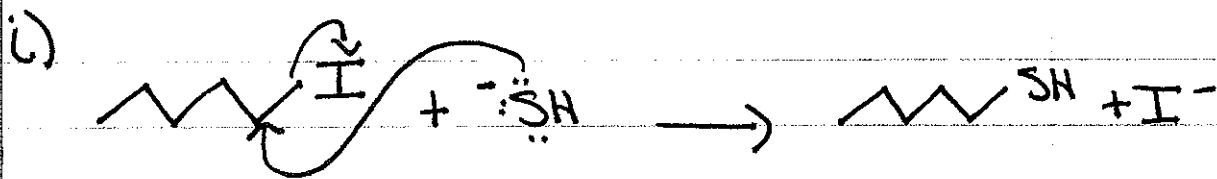
e.) Tosyl chloride + pyridine



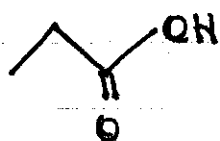
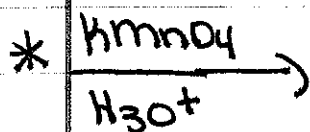
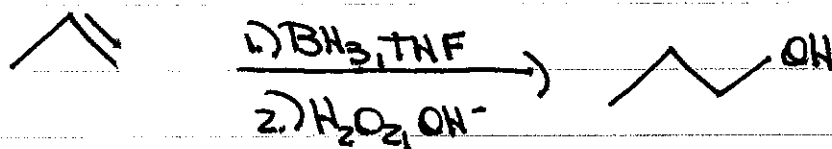
f.)



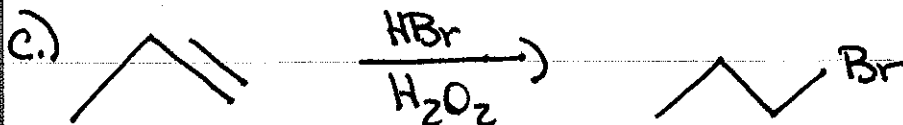
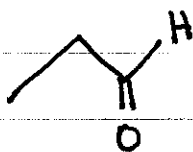
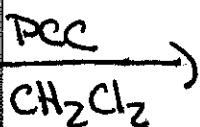
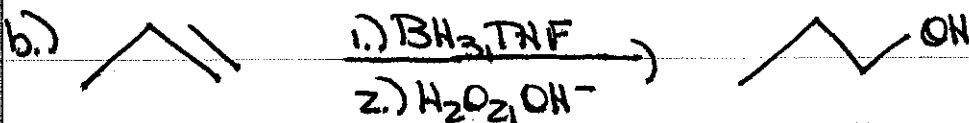


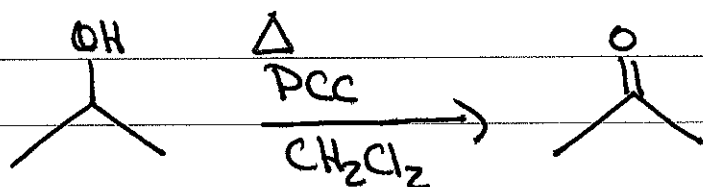
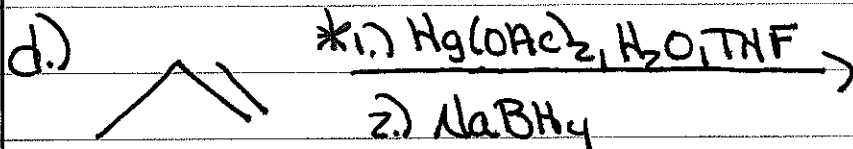


③ a)



* = Could also use Cr





* = $\text{H}_3\text{O}^+/\text{H}_2\text{O}$ could also have been used here
 Δ = $\text{KMnO}_4 + \text{Cr}$ could also have been used here

